Construction of 3,4-Dihydro-1,2-diazete Ring through 4π Electron Cyclization of 4-Hydroxy-2-oxo-2*H* Chromene-3-carbaldehyde [(1*E*)-arylmethylene] Hydrazone

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Published online 31 March 2010 in Wiley InterScience (www.interscience.wiley.com).

A new, short and efficient synthesis of 4-hydroxy-3-(4-aryl-3,4-dihydro-1,2-diazet-3-yl)-2H-chromen-2-one is described in which the 3,4-dihydro-1,2-diazete ring is constructed from arylmethylene hydrazone by 4π electron cyclization as per electrocyclic reaction.

J. Heterocyclic Chem., 47, 513 (2010).

INTRODUCTION

Nitric oxide (NO) has been recognized as an important cellular mediator with diverse biological functions [1,2] including treatment for the respiratory, cardiovascular, infective, and other several diseases [3]. Derivatives of 3,4-dihydro-1,2-diazete-1,2-dioxide have recently been investigated as NO donors in vitro and in vivo and found to be highly effective vasodilators [4]. 1,2-Diazetine N,N-dioxides (diazetine dioxides) are a class of strained four-membered ring azo dioxide heterocycles. Although the first report of a diazetine dioxide was as early as 1971, only a handful of such compounds are currently known [5-9]. Diazetine dioxides have been used as highly effective low-energy triplet quenchers in photochemical reactions [10] and have been recently investigated for their biological activity as potent vasorelaxant and antiaggregant agents [7-9,11]. One of the more intriguing aspects of the reactivity of diazetine dioxides is their tendency to liberate 2 equiv of nitric oxide (NO) upon decomposition to yield the corresponding alkene [5,7]. It is the production of the biologically active molecule NO that has suggested the possibility of using diazetine dioxides as pharmaceutical agents [7,11]. The mechanism by which NO is liberated still remains a question [12,13]. Despite the marked pharmaceutical application of diazetidines [14], isosteric diazetenes [15], 1,2-dihydro-1,2-diazetenes [16], and 3,4dihydro-1,2-dihydro-1,2-diazetenes [17] derivatives, their conjugation with heterocyclic compounds is less studied.

Coumarin and its derivatives represent one of the most active classes of compounds possessing a wide

spectrum of biological activity [18–26]. Many of these compounds have proved to be active as antitumor [18,19], antibacterial [20,21], antifungal [22–24], anticoagulant [25], and anti-inflammatory [26]. In addition, these compounds are used as additives to food and cosmetics [27], dispersed fluorescent and laser [28].

The potential pharmaceutical utility of 1,2-diazete derivatives and 4-hydroxy coumarin derivatives prompted us to synthesize new 3,4-dihydro-1,2-diazete derivatives incorporated with 4-hydroxy coumarin framework 4(a–l).

RESULTS AND DISCUSSION

The synthesis of 4-hydroxy-3-(4-aryl-3,4-dihydro-1,2diazet-3-yl)-2H-chromen-2-one derivatives 4(a-1) is shown in Scheme 1. The precursor 4-hydroxy coumarin 1 was prepared by following the literature methods [29]. A mixture of 1 and triethyl orthoformate containing catalytic amount of p-toluene sulfonic acid (PTSA) was subjected to microwave irradiation at 240 W for 2 min to obtain 3-ethoxymethylene-3*H*-2,4-dione (2) in moderate yield (\sim 60%). The hydrolysis of 2 with K_2CO_3 resulted into 4-hydroxy coumarin-3-carbaldehyde [30]. Subsequent treatment of the 2 with hydrazine hydrate at ambient temperature afforded 4-hydroxy-2-oxo-2H-cromene-3-carbaldehyde- hydrazone (3) in excellent yield $(\sim 90\%)$. It was found that the reaction did not require any solvent or external heating. When hydrazone 3 reacted with various aldehydes at 100°C in DMSO containing con. HCl as a catalyst, 4-hydroxy-3-(4-aryl-3,4dihydro-1,2-diazet-3-yl)-2*H*-chromen-2-one derivatives **Scheme 1.** Reagents and conditions: (a) Triethyl orthoformate. Microwave irradiation at 240 W, 2 min; (b) NH₂NH₂·H₂O, r.t., Stirr.; (c) R-CHO, DMSO, con. HCl, 100°C, 30 min.

4a–l were obtained in good yield (Table 1). One can envisage that the intermediates 4-hydroxy-2-oxo-2H chromene-3-carbaldehyde (arylmethylene) hydrazones underwent 4π electron cyclization to form a four membered 1,2-diazete ring system incorporates with 4-hydroxy coumarin nucleolus **4a–l** (Scheme 1). The formation of 1,2-diazete ring system was confirmed by IR, Mass, 1H NMR, and ^{13}C NMR spectral study.

The IR spectrum of 4a exhibited O—H stretching vibration peak in the range of $3200\text{--}3400~\text{cm}^{-1}$ and C=O Stretching vibration of coumarin at $1690~\text{cm}^{-1}$ indicating the presence of hydroxyl group. The ^1H NMR spectrum of compound 4a displayed the hydroxyl proton at $14.04~\delta$ ppm as a singlet. Two methine protons of diazete ring were observed at $8.8~\text{and}~8.6~\delta$ ppm. The downfield chemical shift of these protons compared with simple 1,2-diazete ring system $(4\text{--}5~\delta)$ ppm reported by G. W. Breton *et al.* [31] and $8.69~\delta$ ppm reported by

Table 1
Physical properties of the synthesized compounds 4(a-l).

Entry	Product	R	Yield (%) ^a	Mp (°C)
1	4a	Ph	75	230-232
2	4b	$4-CH_3-C_6H_4$	71	214-216
3	4c	4-OCH ₃ -C ₆ H ₄	76	206-208
4	4d	3,4-di-OCH ₃ -C ₆ H ₄	85	218-220
5	4e	2,5-di-OCH ₃ -C ₆ H ₄	79	210-212
6	4f	$2-C1-C_6H_4$	65	212-214
7	4g	$4-F-C_6H_4$	65	206-208
8	4h	2-OH-C_6H_4	70	210-212
9	4i	4- NO_2 - C_6H_4	80	214-216
10	4j	4-N,N-di-CH ₃ -C ₆ H ₄	60	204-206
11	4k	3-Pyridyl	62	228-230
12	41	2-Furyl	58	214–216

^a Isolated yields after purification.

Yutaka Ishida *et al.* [32]) may be attributed to their conjugation with coumarin and phenyl ring system attached with the diazete structure. Moreover, the chemical shift of methine proton was also compared with that of compound 3 (7.19 δ ppm) also supporting the formation of diazete framework. The assignment of relative stereochemistry of both methine protons can be carried out based on coupling constant. The larger values of derived coupling constants place both protons in trans position. The mass spectrum of 4a showed three signals M^++1 , M^+ , and M^+-N_2 at 294, 293, and 264, respectively. Extrusion of N_2 from compound in the mass spectra reveals the existence of the cyclic product, which is in accordance with the assigned structure of 4a.

The mechanism involved in the formation of 1,2-diazete ring follows the electrocyclic reaction. Arylmethylene hydrazone is a conjugated chain containing 4π electron system which undergoes 4π electron cyclization as per pericyclic reaction (Scheme 2).

CONCLUSIONS

In summary, novel 4-hydroxy-3-(4-aryl-3,4-dihydro-1,2-diazet-3-yl)-2H-chromen-2-one derivatives (4a–l) were prepared from 4-hydroxy-2-oxo-2H-cromene-3-carbaldehyde hydrazone and various aldehydes in DMSO via electrocyclic reaction through 4π electron cyclization. The pharmacological study of all compounds is currently under investigation.

Scheme 2. Plausible mechanism of 4π electron cyclization.

EXPERIMENTAL

Melting points were determined on electro thermal apparatus using open capillaries and are uncorrected. Thin-layer chromatography was accomplished on 0.2-mm precoated plates of silica gel G60 F_{254} (Merck). Visualization was made with UV light (254 and 365 nm) or with an iodine vapor. IR spectra were recorded on a FTIR-8400 spectrophotometer using DRS prob. ¹H NMR spectra were recorded on a Bruker AVANCE II (400 MHz) spectrometer in DMSO. Chemical shifts are expressed in δ ppm downfield from TMS as an internal standard. Mass spectra were determined using direct inlet probe on a GCMS-QP 2010 mass spectrometer (Shimadzu). Solvents were evaporated with a BUCHI rotary evaporator. All reagents were purchased from Fluka, Sigma Aldrich, Merck, and Rankem and used without further purification.

3-Ethoxymethylene-3*H***-chromene-2,4-dione (2).** A mixture of 6.2 mmol of 4-hydroxycoumarin (1.0 g, 6.2 mmol), triethyl orthoformate (7 mL), and *p*-toluenesulfonic acid monohydrate (0.02 g) was placed in a 50 mL beaker. The beaker was covered with a stem-less funnel and irradiated in the microwave oven for 2 min at 240 W. The resultant residue was cooled to room temperature, the solvent was decanted, and the residue was crystallized in chloroform to give pure yellow crystals. Yield: 0.81 g (60%); mp 140–141°C.

4-Hydroxy-2-oxo-2*H*-cromene-3-carbaldehyde hydrazone (3). 3-(Ethoxymethylene)-3*H*-chromene-2,4-dione (2, 2.18 g, 10 mmol) was stirred at room temperature with excess 50% hydrazine hydrate for about 10–15 min. The solid separated out was filtered and crystallized from chloroform to give 3, yield: 1.83 (90%); mp 138–140°C. IR (KBr): 3528, 3245, 1689, 1545; 1 H NMR (400 MHz, DMSO) δ 8.47 (s, 1H, OH), 7.98 (m, 1H, ArH), 7.55 (m, 1H, ArH), 7.23 (m, 2H, ArH), 7.19 (s, 1H, =CH), 5.88 (s, 2H, NH₂); MS(EI): 204 (M⁺).

General procedure for the synthesis of 4-hydroxy-3-(4phenyl-3,4-dihydro-1,2-diazet-3-yl)-2H-chromen-2-one (4a). A mixture of equimolar amount of 4-hydroxy-2-oxo-2H-chromene-3-carbaldehyde hydrazone 3 and benzaldehyde was dissolved in DMSO containing catalytic amount of con. HCl and heated at 100°C for 30 min with stirring. The reaction mixture was allowed to attain room temperature. The separated solid was filtered off and washed with methanol. The crude product obtained was recrystallized from chloroform to give 4 $hydroxy\hbox{-}3\hbox{-}(4\hbox{-}aryl\hbox{-}3,4\hbox{-}dihydro\hbox{-}1,2\hbox{-}diazet\hbox{-}3\hbox{-}yl)\hbox{-}2H\hbox{-}chromen\hbox{-}2\hbox{-}$ one (4a) as a yellow solid. Yield 75%; mp 230-232°C; IR (KBr) 3200, 1690 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 14.04 (s, 1H, OH), 8.80 (d, J = 12.36 Hz, 1H, CH), 8.67 (d, J =15.52 Hz, 1H, CH), 7.96 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.57 (t, 1H, Ar), 7.45–7.39 (m, 3H, Ar), 7.26–7.17 (m, 2H, Ar); MS(EI) 292 [M]⁺. Anal. Calcd. for C₁₇H₁₂N₂O₃: C, 69.86; H, 4.14; N, 9.58. Found: C, 69.78; H, 4.25; N, 9.45.

Compounds **4b–1** were prepared by following the same procedure as described for **4a**.

4-Hydroxy-3-(4-phenyl-3,4-dihydro-1,2-diazet-3-yl)-2*H***-chromen-2-one (4a).** This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3200, 1690 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.04 (s, 1H, OH), 8.80 (d, J = 12.36 Hz, 1H, CH), 8.67 (d, J = 15.52 Hz, 1H, CH), 7.96 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.57 (t, 1H, Ar), 7.45–7.39 (m, 3H, Ar), 7.26–7.17 (m, 2H, Ar); MS(EI) 292 [M]⁺. Anal. Calcd. for C₁₇H₁₂N₂O₃: C, 69.86; H, 4.14; N, 9.58. Found: C, 69.78; H, 4.25; N, 9.45.

4-Hydroxy-3-[4-(4-methylphenyl)-3,4-dihydro-1,2-diazet-3-yl]-2*H***-chromen-2-one (4b). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3190, 1690 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.14 (s, 1H, OH), 8.95 (d, J = 12.12 Hz, 1H, CH), 8.60 (d, J = 31.64 Hz, 1H, CH), 8.05 (t, 1H, Ar), 7.69 (d, 2H, Ar), 7.67 (t, 1H, Ar), 7.64–7.60 (m, 4H, Ar), 2.42 (s, 3H, CH₃); MS(EI) 306 [M]⁺. Anal. Calcd. for C₁₈H₁₄N₂O₃: C, 70.58; H, 4.61; N, 9.15. Found: C, 70.51; H, 4.54; N, 9.05.**

4-Hydroxy-3-[4-(4-methoxyphenyl)-3,4-dihydro-1,2-diazet-3-yl]-2*H***-chromen-2-one (4c). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3140, 1716 cm⁻¹; ^{1}H NMR (400 MHz, DMSO) \delta 14.07 (s, 1H, OH), 8.86 (d, J=12.48 Hz, 1H, CH), 8.70 (d, J=10.08 Hz, 1H, CH), 8.02 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.62 (t, 1H, Ar), 7.31–7.23 (m, 2H, Ar), 7.00–6.98 (m, 2H, Ar), 3.86 (s, 3H, OCH₃); MS(EI) 322 [M]⁺. Anal. Calcd. for C₁₈H₁₄N₂O₄: C, 67.07; H, 4.38; N, 8.69. Found: C, 67.11; H, 4.24; N, 8.58.**

3-[4-(3,4-Dimethoxyphenyl)-3,4-dihydro-1,2-diazet-3-yl]-4-hydroxy-2*H***-chromen-2-one (4d). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3210, 1699 cm^{-1}; ^{1}H NMR (400 MHz, DMSO) \delta 14.07 (s, 1H, OH), 8.88 (d, J=12.30 Hz, 1H, CH), 8.64 (d, J=14.42 Hz, 1H, CH), 7.99 (t, 1H, Ar), 7.82 (d, 2H, Ar), 7.59 (t, 1H, Ar), 7.48–7.20 (m, 3H, Ar), 3.88 (s, 6H, OCH₃); MS(EI) 352 [M]^{+}. Anal. Calcd. for C₁₉H₁₆N₂O₅: C, 64.77; H, 4.58; N, 7.95. Found: C, 64.61; H, 4.49; N, 7.81.**

3-[4-(2,5-Dimethoxyphenyl)-3,4-dihydro-1,2-diazet-3-yl]-4-hydroxy-2*H***-chromen-2-one (4e). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3235, 1687 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.07 (s, 1H, OH), 8.96 (d, J = 2.64 Hz, 1H, CH), 8.90 (d, J = 12.4 Hz, 1H, CH), 8.03 (t, 1H, Ar), 7.91 (d, 2H, Ar), 7.62 (t, 1H, Ar), 7.44 (t, 1H, Ar), 7.30 (t, 1H, Ar); 7.26 (t, 1H, Ar), 3.88 (s, 6H, OCH₃); MS(EI) 352 [M]⁺. Anal. Calcd. for C₁₉H₁₆N₂O₅: C, 64.77; H, 4.58; N, 7.95. Found: C, 64.59; H, 4.71; N, 7.84.**

3-[4-(2-Chlorophenyl)-3,4-dihydro-1,2-diazet-3-yl]-4-hydroxy- 2H-chromen-2-one (4f). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3221, 1702 cm⁻¹; 1 H NMR (400 MHz, DMSO) δ 14.07 (s, 1H, OH), 8.80 (d, J=12.28 Hz, 1H, CH), 8.68 (d, J=18.06 Hz, 1H, CH), 7.99 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.61 (t, 1H, Ar), 7.43–7.49 (m, 3H, Ar), 7.22–7.27 (m, 1H, Ar); MS(EI) 326 [M] $^{+}$. Anal. Calcd. for C₁₇H₁₁ClN₂O₃: C, 62.49; H, 3.39; N, 8.57. Found: C, 62.37; H, 3.26; N, 8.47.

3-[4-(4-Fluorophenyl)-3,4-dihydro-1,2-diazet-3-yl]-4-hydroxy- 2H-chromen-2-one (4g). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3088, 1716 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.07 (s, 1H, OH), 8.81 (d, J=12.58 Hz, 1H, CH), 8.73 (d, J=10.02 Hz, 1H, CH), 7.99 (t, 1H, Ar), 7.74 (d, 2H, Ar), 7.58 (t, 1H, Ar), 7.44–7.49 (m, 2H, Ar), 7.23–7.26 (m, 2H, Ar); MS(EI) 310 [M]⁺. Anal. Calcd. for C₁₇H₁₁FN₂O₃: C, 65.81; H, 3.57; N, 9.03. Found: C, 65.73; H, 3.51; N, 9.12.

4-Hydroxy-3-[4-(2-hydroxyphenyl)-3,4-dihydro-1,2-diazet-3-yl]-2*H***-chromen-2-one (4h). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3400, 3259, 1696 cm^{-1}; ^{1}H NMR (400 MHz, DMSO) \delta 14.09 (s, 1H, OH), 9.57 (s, 1H, OH), 8.79 (d, J=12.30 Hz,**

1H, CH), 8.66 (d, J=11.38 Hz, 1H, CH), 7.96 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.57 (t, 1H, Ar), 7.46–7.39 (m, 2H, Ar), 7.25–7.17 (m, 2H, Ar); MS(EI) 308 [M]⁺. Anal. Calcd. for $C_{17}H_{12}N_2O_4$: C, 66.23; H, 3.92; N, 9.09. Found: C, 66.12; H, 4.06; N, 8.94.

4-Hydroxy-3-[4-(4-nitrophenyl)-3,4-dihydro-1,2-diazet-3-yl]- 2H-chromen-2-one (**4i**). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3490, 1697 cm⁻¹; 1 H NMR (400 MHz, DMSO) δ 14.07 (s, 1H, OH), 8.81 (d, J=12.18 Hz, 1H, CH), 8.68 (d, J=15.55 Hz, 1H, CH), 7.99 (t, 1H, Ar), 7.73 (d, 2H, Ar), 7.56 (t, 1H, Ar), 7.48–7.57 (m, 3H, Ar), 7.23–7.21 (m, 1H, Ar); MS(EI) 337 [M] $^{+}$. Anal. Calcd. for C₁₇H₁₁N₃O₅: C, 60.54; H, 3.29; N, 12.46. Found: C, 60.47; H, 3.18; N, 12.55.

3-{4-[4-(Dimethylamino)phenyl]-3,4-dihydro-1,2-diazet-3-yl}-4-hydroxy-2*H*-chromen-2-one (4j). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3209, 1719 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.10 (s, 1H, OH), 8.87 (d, J = 12.28 Hz, 1H, CH), 8.71 (d, J = 14.68 Hz, 1H, CH), 8.02 (t, 1H, Ar), 7.75 (d, 2H, Ar), 7.60 (t, 1H, Ar), 7.40–7.32 (m, 2H, Ar), 7.27–7.23 (m, 2H, Ar), 2.97 (s, 6H, N(CH)₂); MS(EI) 335 [M]⁺. Anal. Calcd. for C₁₉H₁₇N₃O₃: C, 68.05; H, 5.11; N, 12.53. Found: C, 67.88; H, 5.03; N, 12.64.

4-Hydroxy-3-(4-pyridin-3-yl-3,4-dihydro-1,2-diazet-3-yl)- 2H-chromen-2-one (**4k**). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3235, 1699 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.10 (s, 1H, OH), 8.80 (d, J = 12.96 Hz, 1H, CH), 8.65 (d, J = 14.93 Hz, 1H, CH), 7.95 (t, 1H, Ar), 7.74 (d, 2H, Ar), 7.58 (t, 1H, Ar), 7.14–6.81 (m, 4H, Pyr.); MS(EI) 293 [M]⁺. Anal. Calcd. for C₁₆H₁₁N₃O₃: C, 65.53; H, 3.78; N, 14.33. Found: C, 65.37; H, 3.59; N, 14.15.

3-[4-(2-Furyl)-3,4-dihydro-1,2-diazet-3-yl]-4-hydroxy-2*H***-chromen-2-one** (4l). This compound was obtained according to the earlier general procedure; Yellow solid; IR (KBr): 3266, 1689 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 14.08 (s, 1H, OH), 8.78 (d, J = 11.08 Hz, 1H, CH), 8.64 (d, J = 15.34 Hz, 1H, CH), 7.90 (t, 1H, Ar), 7.70 (d, 2H, Ar), 7.54 (t, 1H, Ar), 6.98–6.69 (m, 3H, Fur.); MS(EI) 282 [M]⁺. Anal. Calcd. for C₁₅H₁₀N₂O₄: C, 63.83; H, 3.57; N, 9.92. Found: C, 63.78; H, 3.41; N, 9.84.

Acknowledgments. Authors are thankful for facilities and grants given under UGC-SAP for Department Research Support (DRS) and Department of Science and Technology (DST) New Delhi for Fund for Improvement of Science and Technology (FIST) and Department of Chemistry for providing laboratory facilities.

REFERENCES AND NOTES

- [1] Bredt, D. S.; Snyder, S. H. Annu Rev Biochem 1994, 63, 175.
 - [2] Schmit, H.; Walter, U. Cell 1994, 78, 919.
 - [3] Lehmann, J. Expert Opin Ther Pat 2000, 10, 559.
- [4] Utepbergenov, D. I.; Khramtsov, V. V.; Vlassenko, L. P.; Markel, A. L.; Mazhukin, D. G.; Tikhonov, A.Y.; Volodarsky, L. B. Biochem Biophys Res Commun 1995, 214, 1023.
- [5] Singh, P.; Boocock, D. G. B.; Ullman, E. F. Tetrahedron Lett 1971, 42, 3935.

- [6] White, D. K.; Greene, F. D. J Am Chem Soc 1978, 100, 6760.
- [7] Severina, I. S.; Belushkina, N. N.; Grigoryev, N. B. Biochem Mol Biol Int 1994, 33, 957.
- [8] Kirilyuk, I. A.; Utepbergenov, D. I.; Mazhukin, D. G.; Fechner, K.; Mertsch, K.; Khramtsov, V. V.; Blasig, I. E.; Haseloff, R. J Med Chem 1998, 41, 1027.
- [9] Khramtsov, V. V.; Utepbergenov, D. I.; Woldman, Ya. Yu.; Vlassenko, L. P.; Markel, A. L.; Kiriljuk, I. A.; Grigor'ev, I. A.; Mazhukin, D. G.; Tikhonov, A. Ya.; Volodarsky, L. B. Biochemistry (Moscow) 1996, 61, 1223.
- [10] (a) Ullman, E. F.; Singh, P. J. J Am Chem Soc 1972, 94, 5077; (b) Singh, P. J.; Ullman, E. F. J Am Chem Soc 1976, 98, 3018.
- [11] (a) Wang, G. P.; Xian, M.; Tang, X.; Wu, X.; Wen, Z.; Cai, T.; Janczuk, A. J Chem Rev 2002, 102, 1091; (b) Yelinova, V. I.; Bobko, A. A.; Mazhukin, D. G.; Markel, A. L.; Khramtsov, V. V. Russ J Bioorg Chem 2003, 29, 395; (c) Utepbergenov, D. I.; Khramtsov, V. V.; Vlassenko, L. P.; Markel, A. L.; Mazhukin, D. G.; Tikhonov, A. Ya.; Voldarsky, L. B. Biochem Biophys Res Commun 1995, 214, 1023; (d) Severina, I. S.; Ryaposova, I. K.; Volodarsky, L. B.; Mozhuchin, D. C.; Tichonov, A. Ya.; Schwartz, G. Ya.; Granik, V. G.; Grigoryev, D. A.; Grigoryev, N. B. Biochem Mol Biol Int 1993, 30, 357.
- [12] (a) Greene, F. D.; Gilbert, K. E. J Org Chem 1975, 40, 1409; (b) Singh, P. J Org Chem 1975, 40, 1405.
- [13] Snyder, J. P.; Heyman, M. L.; Suciu, E. N. J Org Chem 1975, 40, 1395.
- [14] Zahradnik, M. The Production and Application of Fluorescent Brightening Agent; Wiley, 1992.
- [15] Moore, J. A. In Chemistry of Heterocyclic compounds; Weissberger, A., Ed.; Interscience: New York, 1964; Vol. 19 II, p 916.
- [16] Nunn, E. E.; Warrener, R. N. J Chem Soc Chem Commun 1972, 818.
- [17] Effenberger, F.; Maier, R. Angew Chem Int Ed 1966, 5, 416.
- [18] (a) Raev, L.; Voinov, E.; Ivanov, I.; Popov, D. Pharmazie 1990, 45, 696; (b) Raev, L.; Voinov, E.; Ivanov, I.; Popov, D. Chem Abstr 1990, 114, 74711B.
- [19] Nofal, Z. M.; El-Zahar, M.; Abd El-Karim, S. Molecules 2000, 5, 99.
- [20] El-Agrody, A. M.; Abd El-Latif, M. S.; El-Hady, N. A.; Fakery, A. H.; Bedair, A. H. Molecules 2001, 6, 519.
 - [21] Pratibha, S.; Shreeya, P. Indian J Chem 1999, 38B, 1139.
- [22] Patonay, T.; Litkei, G. Y.; Bognar, R.; Erdei, J.; Misztic, C. Pharmazie 1984, 39, 86.
 - [23] Shaker, R. M. Pharmazie 1996, 51, 148.
 - [24] El-Farargy, A. F. Egypt J Pharm Sci 1991, 32, 625.
- [25] Manolov, I.; Danchev, N. D. Eur J Med Chem Chim Ther 1995, 30, 531.
- [26] Emmanuel-Giota, A. A.; Fylaktakidou, K. C.; Hadjipavlou-Litina, D. J.; Litinas, K. E.; Nicolaides, D. N. J Heterocycl chem 2001, 38, 717.
- [27] Kennedy, R. O.; Thornes, R. D. Counarins: Biology, Applications and Mode of Action; Wiley, 1997.
 - [28] Emeleus, H. J.; Hurst, G. L. J Chem Soc 1962, 3276.
- [29] Stahmann, M. A.; Wolff, I.; Link, K. P. J Am Chem Soc 1943, 65, 2285.
- [30] Rad-Moghadam, K.; Mohseni, M. Monatsh chem 2004, 135, 817.
- [31] Breton, G. W.; Shugart, J. H.; Hughey, C. A.; Perala, S. M.; Hicks, A. D. Org Lett 2001, 3, 3185.
- [32] Ishida, Y.; Donnadieu, B.; Bertrand, G. Proc Natl Acad Sci USA 2006, 103, 13585.